

# Excitation and emission thermal shifts in $\text{ABF}_3\text{:Mn}^{2+}$ perovskites: coupling with impurity vibrational modes

M C Marco de Lucas, F Rodríguez and M Moreno

Departamento de Ciencias de la Tierra y Física Materia Condensada, Facultad de Ciencias, Universidad de Cantabria, 39005 Santander, Spain

Received 13 April 1995, in final form 18 May 1995

**Abstract.** The thermal shifts undergone by the first moment of the  ${}^6\text{A}_{1g}(\text{S}) \rightarrow {}^4\text{T}_{1g}(\text{G})$  excitation band and the associated emission band of  $\text{Mn}^{2+}$ -doped  $\text{ABF}_3$  perovskites are investigated in the 9–300 K temperature range. It is found that these shifts are similar for the whole series and have average values of +150 and +450  $\text{cm}^{-1}$  for excitation and emission, respectively. Both the sign and the magnitude of these *different* thermal shifts are explained in terms of (i) the phonon assistance mechanism required to gain intensity of the parity-forbidden transitions, (ii) the quadratic electron–phonon coupling and (iii) thermal expansion effects. To achieve this analysis a previous discussion upon the nature of the vibrational modes seen in the optical spectra is carried out. It is stressed that the impurity vibrational mode displaying  $\hbar\omega_g = 570 \text{ cm}^{-1}$  in the emission spectrum of  $\text{KMgF}_3\text{:Mn}^{2+}$  exhibits a value of 540  $\text{cm}^{-1}$  in the corresponding excitation spectrum. This situation, which is also found for other modes seen in the optical spectra of  $\text{KMgF}_3\text{:Mn}^{2+}$ , indicates that the mode (though associated with the  $\text{LO}_3$  branch of  $\text{KMgF}_3$ ) is not a *pure* mode of the lattice but displays a kind of resonant character. As a salient feature the calculated thermal shifts are based on the experimental shifts experienced by the frequencies of the optical and acoustic modes on going from the ground  ${}^6\text{A}_{1g}$  to the excited  ${}^4\text{T}_{1g}$  state of  $\text{MnF}_6^{4-}$ . At variance with findings for the *R* lines in  $\text{Cr}^{3+}$  and  $\text{V}^{2+}$ , it is clearly demonstrated that the explicit and implicit contributions to the thermal shift of the zero-phonon line in  $\text{MnF}_6^{4-}$  are similar and both induce red shifts upon heating. Moreover the present analysis reveals that the explicit contribution to the thermal shift undergone by the zero-phonon line of  $\text{KMgF}_3\text{:Mn}^{2+}$  is mainly dominated by the odd-parity low-energy modes. The calculated thermal shifts reproduce reasonably well the experimental data.

## 1. Introduction

The microscopic origin of the thermal shift undergone by crystal field bands of transition metal complexes is still a subject of intensive investigations, which deserves precise analysis in order to clarify the different mechanisms involved in such shifts. Let us designate by  $E(T)$  the temperature dependence of the band maximum. Pioneering theories [1–5] related  $E(T) - E(0)$  to the vibrational energy. So it was proposed that

$$E(T) - E(0) \propto \hbar\omega_{\text{eff}}n_{\text{eff}} \quad \text{or} \quad \propto \int_0^{\omega_p} \hbar\omega\rho(\omega)n(\omega) d\omega$$

depending on whether a local description with one effective vibrational mode of energy  $\hbar\omega_{\text{eff}}$  or an extended phonon state model was considered. Both descriptions are able to explain roughly the thermal band shifts by proper selection of the parameters of the model, which are often obtained by fitting procedures. However, the main problem for a precise understanding of the thermal shifts is to *explain* the values of these parameters in terms of

*microscopic models.* Nowadays, theoretical studies as well as experiments in this field are focused on determining the relevance of the different mechanisms involved in the thermal shift, such as the contribution from thermal expansion effects or from high-order terms of the electron-phonon interaction hamiltonian. If a detailed microscopic characterization of the thermal shifts is not a simple task for sharp peaks like the  $R$  lines of  $\text{Al}_2\text{O}_3:\text{Cr}^{3+}$  [6] or  $\text{MgO}:\text{V}^{2+}$  [7], this characterization is even harder for electric dipole (ED) transitions in centrosymmetric complexes, involving changes of the electronic configuration from the ground state,  $e_g^n t_{2g}^m$ , to other excited ones:  $e_g^{n-1} t_{2g}^{m+1}$  or  $e_g^{n+1} t_{2g}^{m-1}$ . In such cases, the thermal shift is strongly influenced not only by the phonon assistance mechanism required to gain ED oscillator strength but also by the thermal expansion, since the transition energy depends directly on the ligand field strength described by the  $10Dq$  parameter. In particular, this latter aspect is difficult to take into account when dealing with doped materials, unless the local structure around the impurity is known. Apart from a proper characterization of the linear electron-phonon coupling responsible for the phonon sideband, the analysis of thermal shift for this kind of transition requires a precise knowledge of the nature of the odd-parity vibrations involved in the transition mechanism as well as of the variation of the zero-phonon line (ZPL) induced either by the changes of the thermal population at constant volume (explicit contribution) or by changes in  $R$  due to the thermal expansion of the complex (implicit contribution). In the usual cases, this is a complex problem to overcome given that these transitions involve broad absorption or emission bands for which the observation of vibronic fine structure and ZPLs is, in the best case, only possible at low temperatures.

Thermal shifts have been widely investigated for sharp peaks involving spin-flip transitions such as the  ${}^4A_2 \rightarrow {}^2E$  in ruby [4] or  $\text{MgO}:\text{Cr}^{3+}$  [3] and  $\text{V}^{2+}$  [8]. Since these transitions are almost independent of  $10Dq$ , their intensity is mostly concentrated in the ZPL, and the implicit contribution of the thermal shift is much smaller than those for broad bands. Recent estimates of such contributions for the  $R$  lines in  $\text{MgO}:\text{Cr}^{3+}$  [6] and  $\text{V}^{2+}$  [7] indicate that the implicit shift is about  $+5 \text{ cm}^{-1}$  in the 4–300 K range, which must be compared with the total shift of  $-17 \text{ cm}^{-1}$  experienced by the peak in the same temperature range. Thus the explicit contribution represents about the 80% of the thermal shift.

The aim of the present work is to investigate the temperature dependence of the  ${}^6A_{1g} \rightarrow {}^4T_{1g}$  excitation band and the corresponding luminescence band of  $\text{Mn}^{2+}$  impurity placed in the series  $\text{KMgF}_3$ ,  $\text{KZnF}_3$ ,  $\text{RbCdF}_3$ ,  $\text{RbCaF}_3$  and  $\text{CsCaF}_3$  fluoroperovskites. Also spectroscopic data [9] from the pure  $\text{KMnF}_3$  and  $\text{RbMnF}_3$  perovskites are included for comparison purposes. The temperature range where experiments have been carried out is 9–300 K. The emphasis will be placed on explaining the sign and the values of these variations rather than on the shape of their temperature dependence, which nearly follows hyperbolic cotangent functions like  $E(T) = E(0) + A \coth(\hbar\omega_{\text{eff}}/2kT)$  or the corresponding Debye-type function. This characteristic behaviour, which can also account for the thermal dependence of either the vibrational lattice energy,  $E_{\text{vib}}(T)$ , or the variation of interatomic distances by thermal expansion,  $R(T)$ , could explain why some scientists interpret band shifts either through the variations of  $10Dq$  by changes in  $R$  due to thermal expansion effects or mainly by explicit contributions like for the  $\text{V}^{2+}$  or  $\text{Cr}^{3+}$   $R$  lines. The thermal shifts of the first  ${}^4T_{1g} \rightarrow {}^6A_{1g}$  absorption band observed in  $\text{MnCl}_2$  [10] and in  $\text{RbMnF}_3$  and  $\text{KMnF}_3$  [11, 12] were explained in terms of these two extreme behaviours, respectively. Given that, in both cases,  $E(T)$  obeys the same thermal statistics, an adequate interpretation of these shifts necessarily requires a detailed analysis of the different contributions.

The selection of the fluoroperovskite series for this work is due to the following reasons:

(1) The  $\text{MnF}_6^{4-}$  complexes, which are responsible for the electronic spectra, display a

perfect  $O_h$  symmetry at room temperature. This symmetry is kept at low temperatures even in crystals undergoing structural phase transitions like  $RbCdF_3$  ( $T_c = 124$  K) [13] and  $RbCaF_3$  ( $T_{c1} = 193$  K and  $T_{c2} = 40$  K) [14]. In particular, the tetragonal distortion,  $\Delta R = R_{ax} - R_{eq} = 0.011$  Å displayed by the  $CdF_6^{4-}$  octahedra at 30 K, is only 0.002 Å for  $MnF_6^{4-}$  as demonstrated by the analysis of ENDOR data reported by Studzinski and Spaeth [15]. (2) The presence of isolated  $Mn^{2+}$  in the doped crystals prevents migration processes responsible for the extrinsic luminescence in concentrated compounds [5]. (3) The local structure around the  $Mn^{2+}$  is known for the whole series. In particular, the actual Mn–F distance,  $R$ , of the  $MnF_6^{4-}$  unit embedded in the whole fluoroperovskite series has been determined through the analysis of EPR and optical parameters [16]. (4) The temperature dependence of the lattice parameter,  $a(T)$ , is known for different fluoroperovskites [16]. (5) A salient feature concerning the present crystals is the observation of ZPLs in the low-temperature luminescence spectra. This result was important to establish the  $R$  dependence of the ZPL energy, which exhibits a linear behaviour in the 2.06–2.16 Å range, with  $\partial E_{ZPL}/\partial R = +18\,700$  cm $^{-1}$  Å $^{-1}$  [16]. (6) The low-temperature excitation [17] and luminescence [18] spectra of the  $KMgF_3:Mn^{2+}$  crystal display a rich vibronic structure, which provides suitable information about the vibrational frequencies of the coupled modes for the electronic ground state as well as for the  $^4T_{1g}$  excited one. This information becomes *fundamental* to perform quantitative estimations of the explicit contribution to the thermal shifts. (7) We have made precise measurements of the whole excitation and luminescence bands associated with the first  $^4T_{1g}$  state in the 9–300 K temperature range in order to determine thermal shifts from moment analysis. It is also noteworthy that the use of  $Mn^{2+}$  cations is advantageous for exploring thermal shifts because the  $^6A_{1g} \rightarrow ^4T_{1g}$  excitation band is far away from other excited states, at variance with the  $^4A_2 \rightarrow ^4T_2$  band in weak-field  $Cr^{3+}$  complexes, where the presence of Fano resonances makes it difficult to perform such an analysis.

As pointed out in section 3.1, the thermal shifts are connected with the changes experienced by the frequencies of impurity vibrational modes on passing from the ground state to an excited state. Thus because of the importance played by these impurity modes, a discussion about their nature is given in section 3.2.

## 2. Experimental results

Table 1 collects the values of the more relevant optical parameters together with some structural data of the  $ABF_3:Mn^{2+}$  crystal series. It is worth noting that in the present case we give the values of the first moments,  $M_{1em}$  and  $M_{1ex}$ , instead of the band maxima  $E_{em}$  and  $E_{ex}$  reported in previous works [16]. A fundamental reason for this choice is that moments are more directly calculated than band maxima in the theoretical microscopic models. Moreover, its use also avoids errors in the thermal shift measurement when this is made from the band maxima. In particular, these errors can be important if the band shows rich vibrational structure or the bandshape changes with temperature. However, it must be noticed that, in the present case, moments have been calculated directly upon the measured intensity,  $I(\omega)$ , without any instrumental correction. Although this procedure may induce some deviations in the absolute values of the moments, it gives suitable results when we analyse *variations of moments* induced either by temperature or by changing the host crystal. For the present crystals, we estimate the absolute error of the first moment to be about 100 cm $^{-1}$  though the values,  $M_{1em}$  and  $M_{1ex}$ , can be measured with an accuracy of 20 cm $^{-1}$ . The experiments were performed for the whole  $ABF_3:Mn^{2+}$  series employing the

**Table 1.** Experimental values ( $\text{cm}^{-1}$ ) of the emission and excitation first moment,  $M_{\text{em}}$  and  $M_{\text{exc}}$ , and bandwidths,  $H_{\text{exc}}$  and  $H_{\text{em}}$ , of the  ${}^6\text{A}_{1\text{g}}(\text{S}) \leftrightarrow {}^4\text{T}_{1\text{g}}(\text{G})$  transition of  $\text{Mn}^{2+}$  at  $T = 300$  K and low temperature along the  $\text{ABF}_3:\text{Mn}^{2+}$  series. The  $R_{\text{L}}$ ,  $R$  and  $\Delta R_{\text{L}}$  values ( $\text{\AA}$ ) are the host B-F distance, the Mn-F distance and the variation of the B-F distance from low temperature to 300 K, respectively. Zero-phonon energies,  $E_{\text{ZPL}}$ , are taken from [16];  $\tau$  (ms) is the experimental low-temperature lifetime; and  $\hbar\omega_{\text{eff}}(\text{cm}^{-1})$  is the energy of the effective vibrational mode derived by fitting the experimental  $\tau(T)$  values to the equation  $\tau(T) = \tau(0) \tanh(\hbar\omega_{\text{eff}}/2kT)$ . Estimated errors for the moments are  $\pm 20 \text{ cm}^{-1}$ .

| Compound           | $T$ | $R_{\text{L}}$ | $R$   | $\Delta R_{\text{L}}$ | $M_{\text{exc}}$ | $E_{\text{ZPL}}$ | $M_{\text{em}}$ | $H_{\text{ex}}$ | $H_{\text{em}}$ | $\tau$ ( $T = 9$ K) | $\hbar\omega_{\text{eff}}$ |
|--------------------|-----|----------------|-------|-----------------------|------------------|------------------|-----------------|-----------------|-----------------|---------------------|----------------------------|
| KMgF <sub>3</sub>  | 300 | 1.993          | 2.070 | —                     | 18 250           |                  | 16 850          |                 |                 |                     |                            |
|                    | 9   |                |       |                       | 18 100           | 17 222           | 16 400          |                 |                 | 186                 | 220                        |
| KZnF <sub>3</sub>  | 300 | 2.026          | 2.080 | 0.006                 | 18 620           |                  | 17 000          |                 |                 |                     |                            |
|                    | 9   |                |       |                       | 18 500           | 17 509           | 16 600          |                 |                 | 143                 | 260                        |
| KMnF <sub>3</sub>  | 300 | 2.095          | 2.095 | 0.008                 | 18 900           |                  | Non-lumin.      |                 |                 |                     |                            |
|                    | 4.2 |                |       |                       | 18 760           | 17 883           | Extrinsic       |                 |                 |                     |                            |
| RbMnF <sub>3</sub> | 300 | 2.120          | 2.120 | 0.006                 | 19 300           |                  | Non-lumin.      |                 |                 |                     |                            |
|                    | 4.2 |                |       |                       | 19 120           | 18 221           | Extrinsic       |                 |                 |                     |                            |
| RbCdF <sub>3</sub> | 300 | 2.200          | 2.130 | 0.007                 | 19 590           |                  | 17 780          | 1240            | 1760            |                     |                            |
|                    | 9   |                |       |                       | 19 380           | 18 360           | 17 260          | 950             | 1330            | 135                 | 265                        |
| RbCaF <sub>3</sub> | 300 | 2.227          | 2.132 | 0.008                 | 19 600           |                  | 17 790          | 1315            | 1800            |                     |                            |
|                    | 9   |                |       |                       | 19 460           | 18 428           | 17 290          | 1020            | 1390            | 169                 | 310                        |
| CsCaF <sub>3</sub> | 300 | 2.262          | 2.155 | 0.008                 | 20 050           |                  | 18 240          | 1370            | 1830            |                     |                            |
|                    | 9   |                |       |                       | 19 920           | 18 935           | 17 850          | 1060            | 1370            | 156                 | 280                        |

**Table 2.** Excitation and emission thermal shift,  $\Delta E_{\text{ex}}$  and  $\Delta E_{\text{em}}$ , symmetric and antisymmetric parameters,  $\delta_{\text{s}}$  and  $\delta_{\text{a}}$ , at 300 and 9 K and their corresponding variations,  $\Delta\delta_{\text{s}}(300)$  and  $\Delta\delta_{\text{a}}(300)$  (equations (2) and (3)) obtained from the data of table 1 for the  $\text{ABF}_3:\text{Mn}^{2+}$  series. Average values are given in the bottom.  $\hbar\omega_{\text{eff}}^{\text{gr}}$  and  $\hbar\omega_{\text{eff}}^{\text{ex}}$  are the effective phonon energies for the ground and the excited state, respectively, derived from the temperature dependence of the bandwidth (see text). Units in  $\text{cm}^{-1}$ .

| Compound           | $T$ (K) | $\Delta E_{\text{ex}}$ | $\Delta E_{\text{em}}$ | $\delta_{\text{s}}$ | $\delta_{\text{a}}$ | $\Delta\delta_{\text{s}}$ | $\Delta\delta_{\text{a}}$ | $\hbar\omega_{\text{eff}}^{\text{gr}}$ | $\hbar\omega_{\text{eff}}^{\text{ex}}$ |
|--------------------|---------|------------------------|------------------------|---------------------|---------------------|---------------------------|---------------------------|--|--|
| KMgF <sub>3</sub>  | 300     | 150                    | 450                    | 17 550              | 700                 | 300                       | -150                      |  |  |
|                    | 9       |                        |                        | 17 250              | 850                 |                           |                           |  |  |
| KZnF <sub>3</sub>  | 300     | 120                    | 400                    | 17 810              | 810                 | 260                       | -140                      |  |  |
|                    | 9       |                        |                        | 17 550              | 950                 |                           |                           |  |  |
| RbCdF <sub>3</sub> | 300     | 210                    | 520                    | 18 685              | 905                 | 365                       | -155                      | 281                                    | 271                                    |
|                    | 9       |                        |                        | 18 320              | 1060                |                           |                           |  |  |
| RbCaF <sub>3</sub> | 300     | 140                    | 500                    | 18 695              | 905                 | 320                       | -180                      | 290                                    | 287                                    |
|                    | 9       |                        |                        | 18 375              | 1085                |                           |                           |  |  |
| CsCaF <sub>3</sub> | 300     | 130                    | 390                    | 19 145              | 905                 | 260                       | -130                      | 288                                    | 264                                    |
|                    | 9       |                        |                        | 18 885              | 1035                |                           |                           |  |  |
| Average            | 300     | 150                    | 450                    |                     |                     | 300                       | -150                      |  |  |
| $\sigma$           |         | 30                     | 50                     |                     |                     | 40                        | 20                        |  |  |

same procedure as described elsewhere [13, 14, 16]. The results are summarized in tables 1 and 2.

Inspection of table 1 shows that (1) both the excitation and emission bands shift to higher energies upon heating. The corresponding energy shifts,  $\Delta E_i(300) = M_{1i}(300) - M_{1i}(9)$  ( $i = \text{em}$  or  $\text{ex}$ ), for emission are *higher* than for excitation,  $\Delta E_{\text{em}}(T) > \Delta E_{\text{ex}}(T)$ , at any

temperature along the whole series. These shifts are similar for all crystals and have average values  $\Delta E_{em}(300) = 450 \text{ cm}^{-1}$  and  $\Delta E_{ex}(300) = 150 \text{ cm}^{-1}$ , irrespective of  $R$  (table 2). In this way, it must be pointed out that  $R$  mainly influences the transition energy for each crystal but not its thermal shift. (2) The radiative lifetime at low temperature ranges from 135 to 186 ms. Its temperature dependence follows hyperbolic tangent functions characteristic of ED phonon-assisted transitions with effective phonon energies,  $\hbar\omega_{eff} \approx 200\text{--}300 \text{ cm}^{-1}$  [13, 14, 18]. These values are similar to those spectroscopically obtained from the ED peaks at low temperature. (3) The variation of the Mn–F distance,  $\Delta R(T) = R(T) - R(9)$ , at 300 K is taken as  $\Delta R(300) = 0.007 \text{ \AA}$  for all crystals. Although  $\Delta R(300)$  has not been measured at the impurity level, this value is very similar to that found in the pure manganese perovskites  $KMnF_3$  (0.008  $\text{\AA}$ ) and  $RbMnF_3$  (0.006  $\text{\AA}$ ) by x-ray diffraction as well as to the variation of the divalent cation–fluoride distance in the perovskites:  $RbCdF_3$  (0.007  $\text{\AA}$ ),  $RbCaF_3$  (0.008  $\text{\AA}$ ),  $KZnF_3$  (0.006  $\text{\AA}$ ) and  $CsCaF_3$  (0.008  $\text{\AA}$ ) [16]. It must be emphasized that, though we are implicitly assuming the same  $\Delta R$  value for the  $MnF_6^{4-}$  and the host lattice, this procedure would lead to uncertainties in  $\Delta R(300)$  of only about 0.003  $\text{\AA}$  if we consider that the local thermal expansion around the Mn is twice or half the host lattice value. However, the 0.007  $\text{\AA}$  value is expected to be near the actual value if we take into account that the excitation energy shift for doped perovskites and the pure  $RbMnF_3$  and  $KMnF_3$  [9] is nearly the same (tables 1 and 2). This estimate will be important to calculate the implicit contribution to the band shift.

### 3. Analysis and discussion

#### 3.1. Microscopic model

Within a quasiharmonic approximation, the first moment of the excitation and emission bands involving the  ${}^4T_{1g}$  state of  $Mn^{2+}$  is given by [19]

$$M_1(T) = E_{ZPL}(T) \pm \sum S_i \hbar\omega_i(T) \pm \hbar\omega_{eff} \tanh(\hbar\omega_{eff}/2kT) \quad (1)$$

where signs + and – are taken for excitation and emission, respectively;  $E_{ZPL}(T)$  is the ZPL energy at a given temperature;  $S_i$  is the Huang–Rhys factor of the  $i$ th linearly coupled phonon of angular frequency  $\omega_i(T)$ ; and the third term represents the thermally averaged energy of the ED origin upon which the phonon sideband is built. The latter term is written in terms of an effective phonon of energy  $\hbar\omega_{eff}$ , according to the thermal dependence of the lifetime  $\tau(T)$  measured for the doped crystals [13, 14, 16, 18]. The contribution to the Stokes shift by this ED mechanism, which is  $2\hbar\omega_{eff}$  at  $T = 0 \text{ K}$ , becomes zero in the limit  $kT \gg \hbar\omega_{eff}$  and consequently is responsible for the higher shift experienced by the emission band provided that the ZPLs shift to higher energies upon heating. It is worth noticing that, though the  $\omega_i$  frequencies are the same for the ground and the excited state within the linear coupling regime, this situation changes a little when the quadratic terms are considered. Through the present study we shall use equation (1) as a reasonable first approximation. The effects of quadratic coupling will however be considered in connection with the temperature dependence of  $E_{ZPL}$ . Interestingly, the  $Mn^{2+}$  offers the possibility of separating the different contributions of equation (1) since the whole emission and excitation bands can be properly obtained from the optical spectra. In fact, if we define the symmetric,  $\delta_s(T)$ , and antisymmetric,  $\delta_a(T)$ , parameters by  $\delta_s(T) = [M_{1ex}(T) + M_{1em}(T)]/2$  and  $\delta_a(T) = [M_{1ex}(T) - M_{1em}(T)]/2$  then we obtain from equation (1)

$$\delta_s(T) = E_{ZPL}(T) \quad \text{and} \quad \delta_a(T) = \sum S_i \hbar\omega_i(T) + \hbar\omega_{eff} \tanh(\hbar\omega_{eff}/2kT). \quad (2)$$

Within this scheme,  $\delta_s$  and  $\delta_a$  represents the ZPL energy and half the Stokes shift, respectively. The accuracy of such an approximation can be checked by comparing the values of  $\delta_s(T)$  at low temperatures with  $E_{\text{ZPL}}$  measured spectroscopically (tables 1 and 2). It can be noticed that the difference  $\delta_s(9) - E_{\text{ZPL}}$  is always smaller than  $50 \text{ cm}^{-1}$ . Moreover as the uncertainty involved in the experimental value is about  $100 \text{ cm}^{-1}$ , the difference  $\delta_s(9) - E_{\text{ZPL}}$  can be considered as small. Thus the validity of equation (1) for analysing the experimental first moments of the excitation and emission bands is certainly supported by this result. The small deviations between  $\delta_s(9)$  and  $E_{\text{ZPL}}$  found along the series can be associated either with slight deviations of the ground- and excited-state vibrational frequencies or with the lack of instrumental correction. The variations of these parameters from 9 K to  $T$  are given by

$$\begin{aligned}\Delta\delta_s(T) &= \delta_s(T) - \delta_s(9) = \Delta E_{\text{ZPL}}(T) \\ \Delta\delta_a(T) &= \delta_a(T) - \delta_a(9) = \sum \Delta\{S_i\hbar\omega_i(T)\} - 2\hbar\omega_{\text{eff}}/[\exp(\hbar\omega_{\text{eff}}/kT) + 1].\end{aligned}\quad (3)$$

The use of these parameters is important in order to minimize the effects due to variations of the excited- and ground-state phonon energies. In the second equation, we neglect the variation of  $\hbar\omega_{\text{eff}}$  with temperature given that the energy shift for phonons having energies smaller than  $300 \text{ cm}^{-1}$  is less than  $5 \text{ cm}^{-1}$  in the 100–300 K range as measured by IR spectroscopy in the  $\text{RbCaF}_3$ ,  $\text{RbCdF}_3$  and  $\text{KZnF}_3$  fluoroperovskites [20]. Table 2 collects the values of these parameters at 300 K obtained from the experimental data of table 1. It must be observed that, at variance with the moment values,  $M_{1\text{ex}}$  and  $M_{1\text{em}}$ , strongly dependent on  $R$ , the thermal shifts,  $\Delta\delta_s(T)$  and  $\Delta\delta_a(T)$ , are similar for all crystals independent of their  $R$  value and have average values of  $+300$  and  $-150 \text{ cm}^{-1}$  at 300 K, respectively.

Both the sign and the magnitude of  $\Delta\delta_a(300)$  are well explained through equation (3). In fact, the more relevant contribution to  $\Delta\delta_a(300)$  comes from the ED phonon assistance mechanism, which gives a shift of  $-115 \text{ cm}^{-1}$  practically for any value of  $\hbar\omega_{\text{eff}}$  in the  $220\text{--}320 \text{ cm}^{-1}$  range. The remaining  $-35 \text{ cm}^{-1}$  can be accounted for from the first term of equation (3), by considering that the Stokes shift,  $\sum S_i\hbar\omega_i(T)$ , can be described by one effective phonon as  $S\hbar\omega$  using  $S \sim 1\text{--}2$  and  $\hbar\omega \sim 400\text{--}500 \text{ cm}^{-1}$  as typical values for  $\text{MnF}_6^{4-}$  [21]. Then we estimate  $\Delta\{S\hbar\omega\}$  to be between  $-25$  and  $-50 \text{ cm}^{-1}$  taking into account that the energy shift of the high-frequency  $\Gamma_{\text{LO}_2}$  mode measured by IR spectroscopy for  $\text{KZnF}_3$  is  $-25 \text{ cm}^{-1}$  from 100 to 300 K [20], and assuming that the effective Huang–Rhys factor does not change significantly in this temperature range. Therefore both contributions reasonably account for the experimental value  $\Delta\delta_a(300) = -150 \text{ cm}^{-1}$ . This result also confirms the importance of the phonon assistance mechanism for explaining the different thermal shift experienced by the emission and excitation bands associated with ED transitions.

The variation of the ZPL energy with the temperature, i.e.  $\Delta\delta_s(T)$ , is somewhat more difficult to estimate. In fact, taking the ZPL energy as a thermodynamic variable, its temperature dependence at constant pressure contains two main contributions:

$$(\partial E_{\text{ZPL}}/\partial T)_P = (\partial E_{\text{ZPL}}/\partial T)_v + (\partial E_{\text{ZPL}}/\partial R)_T(\partial R/\partial T)_P \quad (4)$$

which are known as the explicit and implicit terms, respectively. In the present case, the implicit contribution can be easily estimated since we know the variation of  $E_{\text{ZPL}}$  with  $R$  and also  $\Delta R(300)$ . Taking the experimental values,  $\partial E_{\text{ZPL}}/\partial R = +18\,700 \text{ cm}^{-1} \text{ \AA}^{-1}$  and  $\Delta R(300) = 0.007 \text{ \AA}$ , we obtain  $\Delta E_{\text{ZPL}}(300) = +130 \text{ cm}^{-1}$ , which represents 43% of the total contribution  $\Delta\delta_s(300)$ , in agreement with previous findings on the first  ${}^6\text{A}_{1g} \rightarrow {}^4\text{T}_{1g}$  absorption band maximum of  $\text{RbMnF}_3$  and  $\text{KMnF}_3$  [9]. This result clearly points out that both the explicit and implicit terms of equation (4) are similar and play an important role

in the thermal band shift for these strongly  $10Dq$ -dependent transitions. Accordingly, the remaining  $170\text{ cm}^{-1}$  must correspond to the explicit term, which should have the same sign as the implicit contribution. This result contrasts with that observed for the  $Cr^{3+}$  and  $V^{2+}$   $R$  lines where the thermal shift of  $-17\text{ cm}^{-1}$  in the 4–300 K range was mostly explained by the explicit contribution in terms of small quadratic couplings revealing ground-state phonon energies slightly higher than those for the excited state [22]. In such a situation, the thermal dependence of the ZPL energy at constant volume is given by

$$E_{ZPL}(T) = E_{ZPL}(0) + \sum (\hbar\omega_i^{ex} - \hbar\omega_i^{gr})n_i \quad (5)$$

where  $E_{ZPL}(0)$  is the ZPL transition energy at 0 K;  $\hbar\omega_i^{ex}$  and  $\hbar\omega_i^{gr}$  are the  $i$ th phonon energy in the excited and ground states, respectively; and  $n_i = [\exp(\hbar\omega_i^{ex}/kT) - 1]^{-1}$  is the corresponding Bose–Einstein occupancy factor. It must be observed that equation (5) has been obtained by considering all transitions between vibronic wavefunctions with the same quantum numbers for the excited and the electronic ground states. Writing the quadratic coupling term as  $\sum B_{kl}Q_kQ_l$ , equation (5) arises from considering the effects of diagonal terms  $B_{kk}$ . The non-diagonal terms  $B_{kl}$  ( $k \neq l$ ) have an important influence upon the second moment and thus the bandwidth but not on the first moment [23, 3, 4]. Note that equation (5) predicts a red or blue shift upon increasing temperature depending on whether the excited-state phonon frequency is smaller than the ground-state one or vice versa. The crucial point for estimating the explicit term is just knowing the value  $\delta\omega_i = \omega_i^{ex} - \omega_i^{gr}$  for each quadratically coupled mode. Although this information is often hidden in the phonon sidebands, we can extract some interesting conclusions analysing the fine peak structure showed by the low-temperature emission and excitation spectra of the  $KMgF_3:Mn^{2+}$  crystal [17, 18]. Before this, it must be noted that the positive sign of this shift in our crystals is a direct consequence of the *hardening* of some vibrations in the  ${}^4T_{1g}$  excited state of  $Mn^{2+}$ .

### 3.2. The nature of vibrational modes seen in the optical spectra

The evaluation of equation (5) requires knowledge of the changes experienced by the vibrational frequencies when a complex like  $MnF_6^{4-}$  goes from the ground to a given excited state. Some partial information about it can be achieved by looking at the vibrational features reflected in excitation and emission spectra. By virtue of this fact it becomes necessary to discuss the *nature* of such features. Sometimes they are associated with internal modes of the complex while in other cases they are assigned to extended modes of the host lattice. To be more specific in the present analysis let us start with the case of an isolated octahedral  $MX_6$  complex. It is well known that, in the optical transition  $|\varphi_g\rangle \rightarrow |\varphi_e\rangle$  between the ground and an excited state, sharp progressions associated with a vibrational mode of the complex,  $Q$ , can appear, provided  $\Gamma_e \times \Gamma_g \supset \Gamma_Q$  is verified. This selection rule means that the symmetric  $a_{1g}$  mode can always give rise to progressions. Moreover if the excited state is an orbital triplet, progressions related to the stretching  $e_g$  and the bending  $t_{2g}$  modes can also appear. Furthermore for forbidden  $d \rightarrow d$  transitions in  $O_h$  symmetry there are also sharp features associated with *odd* local phonons only partially enabling the transition. Such peaks are the ‘false origins’ upon which the progressions (involving even modes) are built. Though  $MX_n$  complexes are observed *embedded* in a given lattice the preceding description involving only internal modes of the complex explains the main experimental features displayed by the low-temperature spectra of some systems. This situation is encountered for instance in the case of  $Cr^{3+}$ -doped  $A_2BMX_6$  elpasolites ( $A, B$  = monovalent cation;  $M$ =trivalent cation) [24–30]. In this lattice the  $MX_6$  octahedra are well separated. At the same time the internal modes of the  $MX_6$  complex become *transverse* lattice modes of the lattice at

$k = 0$ . In other cases the complex appears to be significantly decoupled from the rest of the lattice and thus the main vibrational features observed in optical spectra are again local modes of the  $\text{MX}_n$  complex. An example of this is  $\text{Cr}^{3+}$ -doped fluoroperovskites [31]. In this case the decoupling from the rest of the lattice is favoured by the higher charge of  $\text{Cr}^{3+}$  compared to that of the host cation. Another example of high decoupling corresponds to the  $\text{CuX}_4(\text{NH}_3)_2^{2-}$  centre embedded in  $\text{NH}_4\text{X}$  ( $\text{X} = \text{Cl}, \text{Br}$ ) [32, 33]. In this case where  $\text{Cu}^{2+}$  is located interstitially the Raman spectrum reveals the existence of two sharp  $a_{1g}$  modes associated with the  $\text{Cu-NH}_3$  and  $\text{Cu-X}$  vibrations, respectively. As the frequency of the latter lies in the domain corresponding to host lattice vibrations, it is the fingerprint of a centre substantially decoupled from the rest of the lattice. In the case of divalent transition metal impurities (like  $\text{Mn}^{2+}$ ,  $\text{V}^{2+}$ ,  $\text{Ni}^{2+}$ , ...) doped fluoroperovskite, the vibrational structure seen in optical spectra has been associated with lattice modes of the host lattice [16, 18, 34]. For instance the low-temperature emission spectrum of  $\text{KMgF}_3:\text{Mn}^{2+}$  displays a vibrational progression involving a frequency  $\hbar\omega_g = 570 \text{ cm}^{-1}$  [18]. This value can hardly be related to a local  $a_{1g}$  or  $e_g$  mode of the  $\text{MnF}_6^{4-}$  complex. In fact  $\hbar\omega(a_{1g}) = 220 \text{ cm}^{-1}$  for systems involving isolated  $\text{MnCl}_6^{4-}$  units [35]. Thus taking into account that the replacement of Cl by F as ligand increases the force constant by a factor of 1.7, a figure close to  $\hbar\omega(a_{1g}) = 400 \text{ cm}^{-1}$  for the isolated  $\text{MnF}_6^{4-}$  complex can be expected. Such a value coincides with the estimate of Solomon and McClure [21, 36] from the electronic spectra of  $\text{RbMnF}_3$  and lies between the two highest energy peaks of the infrared spectrum of  $\text{KMgF}_3$  at 478 and  $300 \text{ cm}^{-1}$  [37]. The corresponding  $\text{TO}_3$  and  $\text{TO}_2$  transverse modes at  $k = 0$  ( $\Gamma$  point of the Brillouin zone) involve the stretching of the  $\text{Mg-F}$  bonds [37, 38].

The significant discrepancy between the  $\hbar\omega(a_{1g}) = 400 \text{ cm}^{-1}$  expected for the  $a_{1g}$  mode of the isolated  $\text{MnF}_6^{4-}$  complex and the experimental value  $\hbar\omega_g = 570 \text{ cm}^{-1}$  can however be understood by accepting that the mode seen in the emission spectrum of  $\text{KMgF}_3:\text{Mn}^{2+}$  has a relation with the *longitudinal*  $\text{LO}_3$  branch of the  $\text{KMgF}_3$  lattice [18] where the highest frequencies in the dispersion curve are involved [37]. In particular, the highest frequency of the  $\text{LO}_3$  branch appears at the  $\Gamma$  point of the Brillouin zone, its value being equal to  $\hbar\omega(\text{LO}_3, \Gamma) = 590 \text{ cm}^{-1}$  from the calculations reported by Salaun *et al* [38]. Along the  $\Gamma\text{X}$  direction the frequency variation is relatively smooth; the value corresponding to the X point is equal to  $\hbar\omega(\text{LO}_3, \text{X}) = 547 \text{ cm}^{-1}$ . As regards the local behaviour around the divalent ion, the  $k = 0$  mode of the  $\text{LO}_3$  branch in  $\text{KMgF}_3$  behaves as a combination of the  $a_{1g}$  and  $e_g$  modes of the  $\text{MgF}_6^{4-}$  complex. As is well known a vibrational mode belonging to a longitudinal branch (like the  $\text{LO}_3$  branch of  $\text{KMgF}_3$ ) is in general hardened with respect to the corresponding transverse branch because of the electric field created by the displacement of the whole ions [39]. In the case of  $\text{KMgF}_3$  the  $\text{LO}_3$ - $\text{TO}_3$  energy difference is particularly high in the  $\Gamma\text{X}$  direction [38]. Although the present arguments support a relevant connection between the impurity mode associated with  $\hbar\omega_g = 570 \text{ cm}^{-1}$  (in the emission spectrum of  $\text{KMgF}_3:\text{Mn}^{2+}$  [18]) and the  $\text{LO}_3$  branch of  $\text{KMgF}_3$ , we cannot conclude that such a mode (though extended and with a very similar frequency) is just a pure mode of the lattice. This conforms to the highly localized character of electronic states associated with the impurity, implying that relative changes in the frequency of a given lattice mode would be of the order of  $N_m^{-1}$  ( $N_m$  being the number of lattice cells). The excitation spectrum of  $\text{KMgF}_3:\text{Mn}^{2+}$  [17] reveals however that the progression of  $570 \text{ cm}^{-1}$  in the emission [18] is now of  $540 \text{ cm}^{-1}$ . This *noticeable shift* between both frequencies indicates that the vibrational frequency observed in optical spectra indeed feels the electronic state of the impurity. This situation is of course observed in cases where the electron-phonon coupling



is made with the modes of the complex, but as explained before it is not expected when coupling occurs with an extended mode of the pure lattice. Therefore the only way to understand the experimental results on  $KMgF_3:Mn^{2+}$  is by assuming that the referred mode seen through emission and excitation spectra is extended but its amplitude on the complex is much bigger than that for a normal mode of the host lattice. Designating in a simple way by  $U_c$  and  $U_{r,i}$  the amplitudes on the complex and on the atom  $i$  belonging to the rest of the lattice, then  $|U_{r,i}| \approx 1/N_m$  while  $|U_c|^2$  can be comparable to  $\sum_i |U_{r,i}|^2$ . In other words the vibrational mode involved in excitation and emission spectra of  $KMgF_3:Mn^{2+}$  would be a kind of resonant mode [40] which though *extended* can have a frequency *sensitive* to the electronic state of the  $MnF_6^{4-}$  complex. A tentative model of this impurity mode is depicted in figure 1.

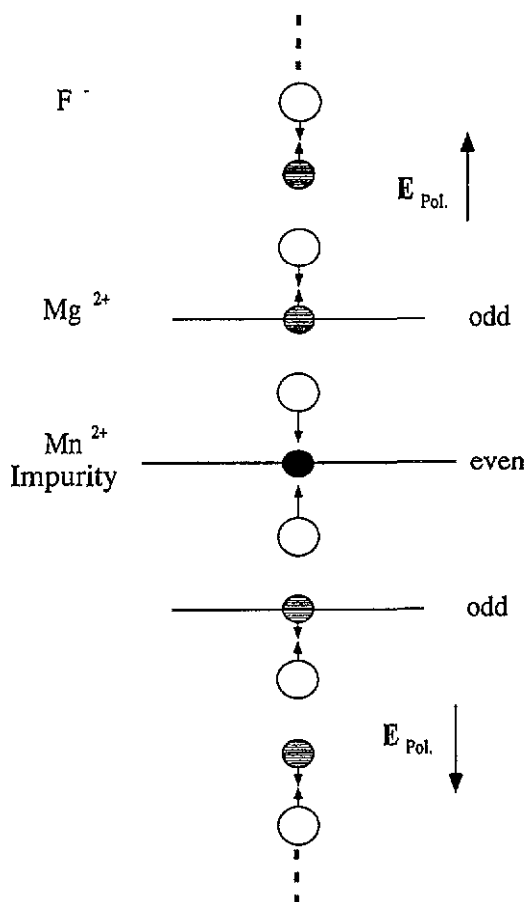


Figure 1. Tentative model of the impurity mode corresponding to  $\hbar\omega_g = 570\text{ cm}^{-1}$  in the emission spectrum of  $KMgF_3:Mn^{2+}$ . Mn is at rest and the two  $F^-$  axial ligands display an even stretching motion around the impurity. The atomic displacements in the upper and lower parts are similar to those corresponding to the  $LO_3(\Gamma)$  phonon of  $KMgF_3$ . Note that the amplitude progressively decreases as far as the lattice ions are more separated from  $Mn^{2+}$ . The dipole moment per unit cell is positive (negative) in the upper (lower) part of the drawing.

The softening of the highest energy mode is also observed in  $RbCdF_3:Mn^{2+}$ ,  $RbCaF_3:Mn^{2+}$  and  $CsCaF_3:Mn^{2+}$  when comparing the emission and excitation bandwidths

as well as their corresponding variation from 9 to 300 K (table 1). The  $T = 9$  K bandwidth is typically about  $1000\text{ cm}^{-1}$  for excitation and  $1300\text{ cm}^{-1}$  in emission, their values increasing 300 and  $430\text{ cm}^{-1}$  from 9 K to 300 K, respectively. Therefore, these data reflect that the effective phonon energies associated with ligand field distortions of  $a_{1g}$  and  $e_g$  symmetry in the electronic ground state are clearly higher than those in the excited state (table 2), according to the spectroscopic observations. These effective frequencies are estimated through the *temperature dependence* of the excitation and emission bandwidths, respectively, by the equation:

$$H(T) = H(0)[\coth(\hbar\omega_{\text{eff}}/2kT)]^{1/2}.$$

The resonant character is also present in other vibrational features seen in the optical spectrum of  $\text{KMgF}_3:\text{Mn}^{2+}$  [17, 18]. As reported in [18] there are peaks in the emission spectrum appearing at 137, 223 and  $367\text{ cm}^{-1}$  from the ZPL. Such peaks have been associated with the  $\text{LA}(X)$ ,  $\Gamma_{25}(\text{t}_{2u})$  and  $\text{LO}_2(\text{t}_{1u})$  lattice modes of  $\text{KMgF}_3$  and appear at 149, 256 and  $401\text{ cm}^{-1}$ , respectively, in the  ${}^6\text{A}_{1g} \rightarrow {}^4\text{T}_{1g}$  excitation band. At this stage, it is worth mentioning that the assignment of the vibrational fine structure and the corresponding vibrational frequencies derived from the luminescence spectrum at 5 K [18] agree with the phonon dispersion curves recently obtained by Salaun *et al* for  $\text{KMgF}_3$  using inelastic neutron scattering and IR data [38].

The resonant character involved in the vibrational modes of  $\text{KMgF}_3:\text{Mn}^{2+}$  seen through optical spectra help us to understand the *sign* of the shift observed in the frequency when we go from the ground to the  ${}^4\text{T}_{1g}$  excited state of  $\text{MnF}_6^{4-}$ . Let us discuss this point in some detail. For transition metal complexes a transition  $t^m e^n \rightarrow t^{m-p} e^{n+p}$  gives rise to a metal–ligand distance,  $R$ , longer in the excited state than in the ground state *provided* that  $p$  is *positive*. A transition of this kind is for instance the first excitation  ${}^4\text{A}_2(\text{t}^3) \rightarrow {}^4\text{T}_2(\text{t}^2\text{e})$  observed for  $\text{CrX}_6^{3-}$  ( $X = \text{halide}$ ) complexes, thus involving the transformation of an antibonding  $\pi$ -electron into an antibonding  $\sigma$ -electron. Such a transformation favours an increase of  $R$  and, at the same time, a softening of the *complex frequencies* so that a given frequency,  $\omega_g$ , in the ground state is higher than  $\omega_e$  corresponding to the excited state. Therefore if the electron–phonon coupling occurs essentially with local modes of the complex, the frequency,  $\omega_g$ , seen in emission spectra is higher than the corresponding one,  $\omega_e$ , detected in the excitation spectra. A good example of this behaviour has been found in cases like  $\text{NH}_4\text{Cl}:\text{Cu}^{2+}$  [32, 33],  $(\text{N-mph})\text{CuCl}_4$  [41] or  $\text{K}_2\text{PtCl}_4$  [42].

In the case of an *isolated*  $\text{MnF}_6^{4-}$  complex the first transition is  ${}^6\text{A}_{1g}(\text{t}^3\text{e}^2) \rightarrow {}^4\text{T}_{1g}(\text{t}^4\text{e})$  and thus in principle  $R$  would be smaller for the excited state. This behaviour is actually obtained in the calculation by Luaña *et al* [43]. These authors find that the difference between  $R_g$  and  $R_e$  (corresponding to the metal–ligand distance for the ground and excited state respectively) is about  $0.07\text{ \AA}$ . Moreover in such a calculation they find that  $\omega({}^6\text{A}_{1g})$  experiences a relative increase of 3% on going from the ground to the  ${}^4\text{T}_{1g}$  state. This idea is confirmed experimentally in the case of  $\text{Na}_3\text{MnCl}_6$  where the  $\text{MnCl}_6^{4-}$  octahedra are well separated and the vibrational modes of the complex also become lattice modes at  $k = 0$ . In this case  $\hbar\omega_g = 214\text{ cm}^{-1}$  while it becomes about 3% higher in the  ${}^4\text{T}_{1g}$  first excited state [35].

Therefore it now becomes clear that the negative value  $\hbar(\omega_e - \omega_g) = -30\text{ cm}^{-1}$  observed for the highest energy mode in  $\text{KMgF}_3:\text{Mn}^{2+}$  cannot be understood on the preceding grounds. By contrast it can be a signature of a resonant mode such as is the value of  $\omega_g$ . It can be thought that the frequency of a resonant mode would depend on the coupling between the vibrations in the complex with those involving ions of the rest of the lattice. Such a coupling would of course be dependent on the electronic state of the complex. Though

this picture provides a reasonable qualitative explanation about the vibrational features seen in  $Mn^{2+}$ -doped fluoroperovskites, additional information is required to settle definitely the present problem.

It is important to realize that odd-parity vibrations of the  $MnF_6^{4-}$  complex such as the  $t_{2u}$  and  $t_{1u}$  modes can be coupled quadratically in spite of the fact that their linear coupling is not symmetry-allowed. In particular, this effect can deeply influence the thermal band shift for temperatures below 300 K if low-energy modes are involved. Elpasolites doped with  $Cr^{3+}$  are clear examples of this behaviour. A different frequency for the ground and excited electronic state of  $Cr^{3+}$  has been observed spectroscopically for odd-parity vibrations of low energy in the  $Cr^{3+}$ -doped  $Cs_2NaInCl_6$  [26] and  $K_2NaGaF_6$  [25] crystals. The vibrational fine structure in the origin of the  ${}^4A_{2g} \rightarrow {}^4T_{2g}$  absorption and luminescence bands indicates that the octahedral  $t_{2u}$  local mode of  $CrCl_6^{3-}$  is the most efficient enabling ED intensity to the transition. The enhanced ED false origins upon which the phonon sideband is made are displaced from the pure magnetic dipole origins  $100\text{ cm}^{-1}$  and  $120\text{ cm}^{-1}$  in absorption and luminescence, respectively (see figures 2 and 3 of [26]). Similar results have also been found in  $K_2NaGaF_6:Cr^{3+}$  from magneto-optical studies [25] where the ground-state  $t_{2u}$  energy,  $\hbar\omega_g = 200\text{ cm}^{-1}$ , is greater than the excited-state one,  $\hbar\omega_e = 160\text{ cm}^{-1}$ . Thus the relevance of these modes to the thermal shift should not be ruled out *a priori*.

The analysis of the optical spectra of  $KMgF_3:Mn^{2+}$  [17, 18] provides useful information about these quadratic couplings. The observed odd-parity  $\Gamma_{25}(t_{2u})$  and  $LO_2(t_{1u})$  energies,  $\hbar\omega_e = 256$  and  $401\text{ cm}^{-1}$  ( $\hbar\omega_g = 223$  and  $367\text{ cm}^{-1}$ ), respectively, clearly indicate that these modes have excited-state frequencies which are neatly higher than the ground-state ones by contrast to the replicas of  $570\text{ cm}^{-1}$  in emission and  $540\text{ cm}^{-1}$  in excitation found for the high-energy even-character mode. It must be noted that the different behaviour exhibited by the low-energy modes of the  $Cr^{3+}$  and  $Mn^{2+}$  complexes in the ground and the first excited states is responsible for the opposite sign of both the explicit and implicit contributions to the thermal shifts. The temperature dependence of the  ${}^4A_2 \rightarrow {}^4T_2$  transition in  $Al_2O_3:Cr^{3+}$  [44] and the present data confirms this behaviour.

### 3.3. Quantitative estimates of the thermal shift

The correlation between a local and a phonon description can be made on the basis that quadratic coupling through phonons is possible whenever the ligand field distortion around the  $Mn^{2+}$  induced by the phonon transforms as one of the quadratically coupled vibrational local modes of  $O_h$  symmetry. If we denote by  $\delta\omega_i = \omega_{ie} - \omega_{ig}$  the difference between the excited- and ground-state quadratic couplings for the  $i$ th phonon mode of a given optical branch, then the contribution to the ZPL thermal shift will be

$$\Delta E_{ZPL} = \sum_i n_i \hbar \delta\omega_i = c \sum_i n_i \hbar \omega_i$$

provided that  $\delta\omega_i/\omega_i = c$  along the  $N_m$  phonons of the branch. For degenerate phonons or slight phonon dispersion, this contribution equals  $cN_m(n_i\hbar\omega_i)$  and therefore would be similar to that corresponding to one strongly coupled resonant phonon of the same frequency whenever the ground state to excited state frequency difference,  $\delta\omega = \omega_e - \omega_g$ , was similar to  $cN_m\omega$ . It means that the contribution to the ZPL thermal shift from  $N_m$  weakly coupled phonons is similar to that of one resonant mode provided that its quadratic coupling spreads out over the  $N_m$  phonons of the optical or acoustic branch.

Consequently, the thermal dependence of the ZPL energy has a very simple analytical equation either for resonant modes or in a pure phonon description assuming *weak* dispersion

within the optical branches and the commonly used approximation,  $\delta\omega_i/\omega_i = c_i$ ; now  $i$  represents the  $i$ th phonon branch. So

$$\Delta E_{\text{ZPL}} = 3c \int_0^{\omega_D} \hbar\omega\rho(\omega)n(\omega) d\omega + 3kT \sum_i c_i \frac{x_i}{\exp(x_i) - 1} \quad (6)$$

where  $c = \delta\omega/\omega$  for the acoustic phonons;  $\rho(\omega)$  is the phonon state density;  $n(\omega)$  is the Bose-Einstein factor;  $\omega$  and  $\omega_D$  are the phonon and Debye angular frequencies, respectively; the factor 3 accounts for the longitudinal and the two transverse modes; and  $x_i = \hbar\omega_i/kT$ . The two terms of equation (6) represent the contribution from the acoustic phonons and the resonant modes (or optical branches), respectively. Note that the acoustic phonons play an important role in the thermal shift due to their low frequencies. The first term of equation (6) is basically the same as employed elsewhere [2-4] for explaining the thermal shifts of the  $\text{Cr}^{3+}R$  lines.

For  $\text{MnF}_6^{4-}$  complexes in fluoroperovskites,  $\Delta E_{\text{ZPL}}$  can be evaluated through equation (6) taking into account that there are 15 vibrational modes in the perovskite structure; three of them spread over the acoustic branches, while the remainder are resonant modes in different optical branches. Although a precise calculation of  $\Delta E_{\text{ZPL}}(300)$  requires knowledge of  $\omega_i$  and  $c_i = \delta\omega_i/\omega_i$  for each mode, its final value must not change significantly for small deviations of the actual  $\omega_i$  and  $c_i$  values.

Therefore, we can estimate the explicit contribution to  $\Delta E_{\text{ZPL}}(300)$  from the vibrational frequencies observed in  $\text{KMgF}_3:\text{Mn}^{2+}$  taking values of  $\hbar\omega$  and  $\delta\hbar\omega$  for the three  $t_{1u} + t_{2u}$  resonant modes of  $\hbar\omega = 570 \text{ cm}^{-1}$  ( $\delta\hbar\omega = -30 \text{ cm}^{-1}$ ) for  $t_{1u}^3$ ,  $367 \text{ cm}^{-1}$  ( $+34 \text{ cm}^{-1}$ ) for  $t_{1u}^2$ ,  $137 \text{ cm}^{-1}$  ( $+12 \text{ cm}^{-1}$ ) for  $t_{1u}^1$  and  $223 \text{ cm}^{-1}$  ( $+33 \text{ cm}^{-1}$ ) for  $t_{2u}$ . From the second term of equation (6), we finally obtain  $\Delta E_{\text{ZPL}}(300) = +105 \text{ cm}^{-1}$  due to the resonant modes. It must be pointed out that the contribution from the *lowest* energy modes,  $t_{1u}^1 + t_{2u}$ , is *almost* 90% of this shift, thus indicating that despite the high-energy modes being the most important for the band shape, their contribution to  $\Delta E_{\text{ZPL}}(T)$  is irrelevant for  $T \leq 300 \text{ K}$ . This  $105 \text{ cm}^{-1}$  value corresponds to 60% of the total experimental explicit shift. Analogously, we calculate the contribution from the acoustic phonons (first term of equation (6)) using the Debye approximation:

$$\Delta E_{\text{ZPL}}(T) = 3c \int_0^{\omega_D} \hbar\omega\rho(\omega)n(\omega) d\omega = 9cN_m\hbar\omega_D \left(\frac{T}{T_D}\right)^4 \int_0^{T_D/T} \frac{x^3}{e^x - 1} dx \quad (7)$$

where  $T_D = (\hbar\omega_D/k)$  is the Debye temperature and  $x = \hbar\omega/kT$ . We estimate  $\omega_D = 4.6 \times 10^{13} \text{ s}^{-1}$  ( $T_D = 350 \text{ K}$ ) for  $\text{KMgF}_3$  through the formula  $\omega_D = (6\pi^2)^{1/3}v/a$  using values  $v = 4.7 \times 10^5 \text{ cm s}^{-1}$  and  $a = 3.986 \text{ \AA}$ . The sound velocity,  $v$ , employed here actually corresponds to the quadratic average of the longitudinal and transverse sound velocities,  $v = [(v_l^2 + 2v_t^2)/3]^{1/2}$ , obtained through either the elastic constants,  $v = [(C_{11} + 2C_{44})/3d]^{1/2}$  ( $C_{11} = 1320 \text{ kbar}$ ,  $C_{44} = 485 \text{ kbar}$  and the density  $d = 3.2 \text{ g cm}^{-3}$ ) [38] or their corresponding sound velocities,  $v_l = 5.8 \times 10^5 \text{ cm s}^{-1}$  and  $v_t = 3.9 \times 10^5 \text{ cm s}^{-1}$  obtained from the phonon dispersion curves [38]. If we take  $N_m\delta\omega/\omega = 0.15$ , according to the  $\delta\omega/\omega$  values measured for the low-frequency modes through the optical spectra, the thermal shift at 300 K from the acoustic modes is  $\Delta E_{\text{ZPL}}(300) = +60 \text{ cm}^{-1}$ . Similar values are obtained using the elastic constants and the lattice parameters of other  $\text{ABF}_3:\text{Mn}^{2+}$  crystals. This value, together with the  $105 \text{ cm}^{-1}$  explicit shift from the resonant modes and the  $120 \text{ cm}^{-1}$  implicit shift, accounts for a total shift of  $+285 \text{ cm}^{-1}$ , which is in good agreement with the experimental value,  $\Delta E_{\text{ZPL}}(300) = 300 \text{ cm}^{-1}$ .

#### 4. Conclusions

The present results clearly point out that the thermal shifts experienced by the first excitation and luminescence bands in  $\text{MnF}_6^{4-}$  complexes are strongly influenced by the ED transition mechanism and both the implicit and the explicit contributions of the ZPL shift. The ED mechanism is mainly responsible for the *different* thermal shift undergone by the excitation and luminescence bands. It has the same sign as the ZPL for emission while the opposite occurs for the excitation band, thus explaining the higher thermal shift found for emission ( $450\text{ cm}^{-1}$ ) than for excitation ( $150\text{ cm}^{-1}$ ).

The implicit contribution is about 40% of the total ZPL shift whereas the remaining 60% is mostly explained in terms of quadratic couplings by the low-energy modes. In particular, it is worth noticing the relevance of the odd-parity modes to the explicit shift despite the fact that they are not linearly coupled to the electronic state. Spectroscopic findings on  $\text{Cr}^{3+}$  complexes also suggest that the deviations of the ground- and excited-state frequencies of the octahedral  $t_{2g}$  modes ( $\delta\omega/\omega \sim 0.2$ ) could be mainly responsible for the opposite sign of the thermal shift exhibited by the  ${}^4A_{2g} \rightarrow {}^4T_{2g}$  in these complexes. A precise experimental determination of both the implicit and explicit contributions in  $\text{CrF}_6^{3-}$  is currently under way. Finally through the present work the different nature of impurity vibrational modes has been discussed. At variance with findings in systems like  $\text{Cr}^{3+}$  in elpasolite and perovskite lattices where the impurity modes are essentially local modes, it is pointed out that in  $\text{Mn}^{2+}$ -doped perovskites a distinct situation comes out. As an important conclusion, the analysis carried out here indicates that in  $\text{ABF}_3\text{:Mn}^{2+}$  the impurity modes observed through the emission and excitation spectra are not exactly modes of the pure lattice but display a kind of resonant character. Further work to improve our knowledge about these impurity modes is necessary.

#### Acknowledgments

Information about the vibrational modes of  $\text{KMgF}_3$  kindly supplied by Professor M Rousseau is acknowledged. This work has been supported by the CICYT (Project No PB92-0505).

#### References

- [1] Engleman R 1960 *Mol. Phys.* **3** 23
- [2] McCumber M D and Sturge D E 1963 *J. Appl. Phys.* **34** 1682
- [3] Imbusch G F, Yen W M, Schawlow A L, McCumber D E and Sturge M D 1964 *Phys. Rev. A* **133** 1029
- [4] Di Bartolo B 1968 *Optical Interactions in Solids* (New York: Wiley)
- [5] Henderson B and Imbusch G F 1989 *Optical Spectroscopy of Inorganic Solids* (New York: Oxford University Press)
- [6] Dong-ping M, Xiao-yi H, Ju-rong C, Ji-ping Z and Zheng-gang Z 1993 *Phys. Rev. B* **48** 4302
- [7] Dong-ping M, Xiao-yi H, Ju-rong C, Yan-young L and Ji-ping Z 1993 *Phys. Rev. B* **48** 14067
- [8] Di Bartolo B and Peccei R 1965 *Phys. Rev. A* **137** 1770
- [9] Rodríguez F, Moreno M, Dance J M and Tressaud A 1989 *Solid State Commun.* **69** 67
- [10] Ronda C R, Siekman H H and Haas C 1987 *Physica B* **144** 331
- [11] Darwish S, Abumansoor S and Seehra M S 1986 *Phys. Rev. B* **34** 3198
- [12] Darwish S and Seehra M S 1988 *Phys. Rev. B* **37** 3422
- [13] Marco de Lucas M C, Rodríguez F, Moreno M and Tressaud A 1994 *J. Phys.: Condens. Matter* **6** 6353
- [14] Marco de Lucas M C, Rodríguez F and Moreno M 1993 *J. Phys.: Condens. Matter* **5** 1437
- [15] Studzinski P and Spaeth J M 1986 *J. Phys. C: Solid State Phys.* **19** 6441
- [16] Marco de Lucas M C, Rodríguez F and Moreno M 1994 *Phys. Rev. B* **50** 2760 and references therein
- [17] Ferguson J, Güdel H U, Krausz E R and Guggenheim H J 1974 *Mol. Phys.* **28** 879
- [18] Rodríguez F, Riesen H and Güdel H U 1991 *J. Lumin.* **50** 101

- [19] Jessop P E and Szabo A 1980 *Opt. Commun.* **33** 301
- [20] Ridou C, Rousseau M and Gervais F 1986 *J. Phys. C: Solid State Phys.* **19** 5757
- [21] Solomon E I and McClure D S 1974 *Phys. Rev. B* **9** 4690
- [22] Nelson D F and Sturge M D 1965 *Phys. Rev. A* **137** 1117
- [23] Henry C and Slichter C P 1968 *Physics of Color Centers* (New York: Academic) p 351
- [24] Greenough P and Paulusz A G 1979 *J. Chem. Phys.* **70** 1967
- [25] Dubicki L, Ferguson J and Von Oosterhout B 1980 *J. Phys. C: Solid State Phys.* **13** 2791
- [26] Güdel H U and Snellgrove T R 1978 *Inorg. Chem.* **17** 1617
- [27] Dolan J F, Kappers L and Bartram R H 1986 *Phys. Rev. B* **33** 7339
- [28] Wood A M, Sinkovits, R S, Charpie J C, Huang W L, Bartram R H and Rossi A R 1993 *J. Phys. Chem. Solids* **54** 543
- [29] Marco de Lucas M C, Rodríguez F, Moreno M and Tressaud A 1991 *J. Lumin.* **48 & 49** 553
- [30] Duclos S J, Vohra Y K and Ruoff A L 1990 *Phys. Rev. B* **41** 5372
- [31] Villacampa B, Casas González J, Alcalá R and Alonso P J 1991 *J. Phys.: Condens. Matter* **3** 8281
- [32] Breñosa A G, Moreno M, Rodríguez F and Couzi M 1991 *Phys. Rev. B* **44** 9859
- [33] Couzi M, Moreno M and Breñosa A G 1994 *Solid State Commun.* **91** 481
- [34] Sturge M D 1971 *Solid State Commun.* **9** 899
- [35] Marco de Lucas M C, Rodríguez F and Moreno M 1994 *Phys. Status Solidi b* **184** 247 and references therein
- [36] Chen M Y, McClure D S and Solomon E I 1972 *Phys. Rev. B* **6** 1690
- [37] Nakagawa I, Tsuchida A and Shimanouchi T 1967 *J. Chem. Phys.* **47** 982
- [38] Salaun S, Mortier M, Gesland J Y, Rousseau M and Hennion B 1993 *J. Phys.: Condens. Matter* **5** 7615
- [39] Brüesch P 1982 *Phonons: Theory and Experiments* (Berlin: Springer)
- [40] Barker A and Sievers A J 1975 *Rev. Mod. Phys.* **47** 51
- [41] McDonald R G and Hitchman M A 1986 *Inorg. Chem.* **25** 3273
- [42] Martin D S, Tucker M A and Kassman A J 1966 *Inorg. Chem.* **4** 1682
- [43] Luafía V, Bermejo M, Flórez M, Recio J M and Pueyo L 1989 *J. Chem. Phys.* **90** 6409
- [44] McClure D S 1962 *J. Chem. Phys.* **36** 2757